# Exam Quantum Engineering (FKA132), Chalmers 2017-08-15

Time and place: 8:00-12:00 on August 15, 2017, Campus Johanneberg. Examinator: Per Hyldgaard (phone 031 772 8422).

Allowed references: Dictionaries, Beta, Physics Handbook, one handwritten paper of size A4 (both sides) with own notes, and a Chalmers-approved calculator. You can also use the "Extra resources" given at the end of this document.

Correct solution of each problem gives 6 points. 15 points are needed to pass. MOTI-VATE YOUR ANSWERS, answers lacking reasonable motivation will not yield full credit.

Problem 1. [Six points total]

a) Write down the time-independent Schrödinger equation and explain all terms of the equation. Assume that the time-independent Schrödinger equation has the set of eigenvalues  $\{E_{\alpha}\}$  and eigenvectors  $\{\psi_{\alpha}\}$ . Write down the general solution  $\psi(\mathbf{r}, t)$  to corresponding time-dependent Schrödinger equation, i.e., the solution to the quantum problem.

b) Oligoacenes are molecules in which multiple benzene rings are stringed together, one after another. Consider two examples, benzene  $C_6H_6$  and anthracene  $C_{14}H_{10}$  and assume that each C atom has a mass equal to 12 times the mass of a proton. Here we consider the dynamics of individual molecules, as dilute gases. If both types molecules are moving at kinetic energy given by the anthracene boiling point,  $K_BT \approx 50$  meV, what are the de Broglie wavelengths of the benzenes and anthracenes? [1 point]

c) Assume that  $a^{\dagger}$  and a are creation and annihiliation operators that, for example, describe a harmonic-oscillator system and satisfy  $[a, a^{\dagger}] = 1$ . Consider next an linear combination  $d = \alpha a + \beta a^{\dagger}$  and evaluate  $[d, d^{\dagger}]$ . [1 point]

d) An harmonic oscillator (describing dynamics in direction x) is described in terms the annihilation operator

$$\hat{a} = \frac{1}{\sqrt{2}} \left( \hat{x} + i \frac{\hat{p}_x}{\hbar} \right),\tag{1}$$

and the Hermitian conjugate creation operators  $\hat{a}^{\dagger}$ . Express the pertubation  $\hat{H}_1(t) = C\hat{x}^2 \cos^2(\omega t)$  in terms of  $\hat{a}$  and  $\hat{a}^{\dagger}$ .

e) Define the concept "electronegativity".

f) State the requirements needed to be fulfilled by a compound to be aromatic (not in the context of having an aroma but in the context of chemical bonding).

#### Problem 2. [6 points total]

A molecule of mass M moves in the potential

$$V(x) = V_0 \left\{ \tanh^2 \left(\frac{x}{a}\right) - 1 \right\}$$
(2)

where  $V_0$  and a are positive constants.

a) What is the "best fit" harmonic oscillator approximation to this potential? That is, consider Taylor-expansion around the energy minimum and express the approximate harmonic-oscillator frequency  $\Omega_0$  in terms of  $V_0$ , a, and the particle mass M.

b) What is the energy spectrum of the harmonic oscillator fit?

The set of normalized steady-state solutions  $|n\rangle$  of the actual potential problem has energies  $\epsilon_n$  (you do not have to compute these).

c) Sketch the wavefuncions for both the lowest- and second-lowest stationary states, assuming your harmonic-oscillator approximation is valid.

d) Sketch both the actual and the model potential, and discuss the quantitative differences that exists between the potentials when  $|x| \gg a$ .

e) For given values of a,  $V_0$ , and M, up to what values of n (number in spectrum) is the harmonic oscillator spectrum a reasonable approximation?

f) Given the state  $|\psi(t=0)\rangle = \frac{1}{2}|2\rangle + \frac{1}{\sqrt{2}}|3\rangle + \frac{1}{C}|5\rangle$  at time t = 0. Determine C and describe the state change at t > 0 (write down  $|\psi(t)\rangle$  for t > 0).

Problem 3. [6 points total]



Figure 1: Sketch of the potential V(x) in problem 3.

A particle of mass m moves in a one-dimensional potential well given by

$$V(x) = \begin{cases} \infty & \text{for } x < -b \\ \eta\left(\frac{x}{b}\right) & \text{for } -b < x < b \\ \infty & \text{for } b < x \end{cases}$$
(3)

where  $\eta < 1$  is a small energy. The potential is illustrated in Fig. 1.

a) Argue that it is useful to split the problem into an unperturbed starting system  $\hat{H}_0$  and a perturbation term  $\hat{H}_1 \equiv \eta(x/b)$ , when you can also assume that  $\eta$  is small compared to the ground-state energy.

b) Calculate the eigenenergy for the lowest state of the system given above to first order in the perturbation  $\hat{H}_1$ .

It may be useful to use the 'Extra resources" at the end.

c) Compute the energy shift of the ground state to second order, assuming that we can focus exclusively on the contribution from the coupling between the lowest two eigenstates.

It may be useful to use the 'Extra resources" at the end.

d) Assume that we irradiate the system at a frequency  $\omega$  so that  $\hbar\omega$  is close to the energy difference between the first and second level,  $\hbar\omega \approx E_2 - E_1$ ; this irradiation is described by the time-dependent perturbation

$$\delta \hat{H}(t) = (e E_{\rm rad}^0) \hat{x} \cos(\omega t). \tag{4}$$

Provide a formal Fermis golden-rule estimate of the rate of transition from the ground to first excited state.

#### Problem 4. [6 points total]

Consider electrons in a homogeneous magnetic field  $\mathbf{B} = B_0(1, 0, 0)$ , where (1, 0, 0) denotes a vector in the x direction. We focus on the spin Hamitonian

$$\hat{H} = -\frac{1}{\mu_0} \boldsymbol{\mu} \cdot \mathbf{B}, \qquad \boldsymbol{\mu} = -\frac{g\mu_B}{2} \boldsymbol{\sigma}$$
(5)

where  $\mu_0$  is the magnetic permability,  $\mu_B = e\hbar/(2m_ec)$  is Bohr magneton,  $g \approx 2.00$ , and where  $\sigma$  represents the Pauli spin matrices.

a) Calculate the eigenvalues and the (normalized) eigenvectors of the hamiltonian  $\hat{H}$ , expressed in terms of electon spinors.

b) At time t = 0 the electron is in its spin-up state, relative to the z-axis. Describe the time evolution of the spin state of the electron at t > 0, i.e., the spin wavefunction as a function of time,  $\chi(t)$ .

c) Calculate the probabilities of finding the electron in the spin-down state,  $P_{\downarrow}(t)$ , (seen relative to direction z), as a function of the time t > 0.

d) Consider finally the dynamics of two electrons moving under the influence of the spin Hamiltonian  $\hat{H}$ . Assume that the electrons occupy the same eigenstate of the Hamiltonian for the spatial wavefunction and ignore all physical interactions between the electrons. If one electron is measured to be in a spin-down configuration at  $t = t_1$ , what is the probability distribution for finding the other electron in spin up or spin down configurations at time  $t_2 > t_1$ ?

## Problem 5. [6 points total]

Fluoride gas has the molecular formula F2.

a) Draw a correct Lewis structure for this compound. [1 point]

b) Construct and draw a molecular orbital energy level diagram for F2 (the interaction between the two 2p-orbitals along the bond axis is much larger than between those perpendicular to this axis). [2 points]

c) For each energy level, give a simple (omit constants) expression for the linear combination of atomic orbitals (AOs) that gives the best approximation of the corresponding molecular orbital (MO). [1 point]

d) Correlate each bonding electron pair and free electron pair, if any, shown in your Lewis structure for F2 with the correct linear combination of atomic orbitals (LCAOs) constructed in question c. [2 points]

### Extra resources

You may find it useful to know the following indefinite-integral results

$$\sin(2y) = 2\sin(y)\cos(y)$$

$$\int \sin^2(y) \, dy = \frac{y}{2} - \frac{\sin(2y)}{4y}$$

$$\int \cos^2(y) \, dy = \frac{y}{2} + \frac{\sin(2y)}{4y}$$

$$\int y \sin^2(y) \, dy = \frac{y^2}{4} - \frac{y}{4}\sin(2y) - \frac{1}{8}\cos(2y).$$

$$\int y \cos^2(y) \, dy = \frac{y^2}{4} + \frac{y}{4}\sin(2y) + \frac{1}{8}\cos(2y).$$

$$\int y \cos(y) \sin^2(y) \, dy = \frac{1}{9} \left[ -3y \cos^3(y) + \cos^2(y) \sin(y) + 2\sin(y) \right].$$

END OF DOCUMENT

FKA 132 resit exam 2017-12-30 Problem 1/ a) Schrödinger equation (SE) in 影子(ちと) = H P(r, と) is given by Hamiltonian  $H = -\frac{42}{5m} \nabla_F^2 + V(F, E)$ where first term is kinetic everyy and second tem is a potential Time-independent Schrödinger equation is (mostly) relevant for timeindependent potential VCF), and given as the eigenvalue problem  $FT \mathcal{V}_{\alpha}(F) = E_{\alpha} \mathcal{V}_{\alpha}(F)$ For a given renergy eigenvector 7acr) the actual (time-dependent SE has the solution Valr, E) = YalF)e - Ext/h.

b) Oliquacenesus Benzene Benzene has mass M=78 mp 2 Where proton Autracene has mass H=178 mp ) 2000 times electron mass me Boiling point = ERIN & SOMEV (~600K) Molecule momentum P=72m ERIN Note (me. Ry) = K = h Plao = 212Traoj ao Bohr radie N. 0.529177 A. de Braghie wave levath X= p= mersy're F. M. Ekin =2TTao/1/36,2 Antracene Benzepie has de Bioglie waveleugth ~ 0,14 À Antraceme has de Broglie verreleyfl~0.09 Å d= aa+bat means dt= atat+bta  $Ed_{d}d^{\dagger}] = \alpha [a, a^{\dagger}] \alpha^{\star} + \beta [a^{\dagger}, a^{\dagger}] \beta^{\star} = |\alpha|^{2} - |\beta|^{2}$ 

 $a = \beta (a + i \frac{p_{x}}{p})$ at= 」(パー: 上) so (a+a+)= 12 x and  $\chi^2 = \frac{1}{2} \left( a + a^{\dagger} \right)^2$ Thus perhabition H= Scose(we) (a+a) e)+ f) See separate solutions for chemistry problems. Problem 2 Potential Vers = Vo 3 tauh 2 (2)-15 touhyphis a good linear behavior inside -0.5 Ky < 0.5 and a fair linianty inside - TET STATE IN HUS restime we may expaud tanh(à) -1 2 (à)

a) Best Rit harmonic potential is thus (X) = Vn(X) = = K·X=V; K= 2V0/a2 Approximate-potential trequency is 20= K/M = 1242 b) Evergy spectroum (as 1D hamonic oscillator En= h 20 (n+ 1/2), n=0,1,---Carlo Yhtx> 3. W<sub>O</sub> X Vn vs Vo V.Vn N Encrases forever Ô Vin Saturates  $=\sqrt{0}$ ×la 421 

S e) Model breakdown when energy eigenvalue En=th 20 (n+1/2) exceeds Vo (tobeep it simple) Thus conteria is triko (2n+1) <26 (=)  $n \leq \frac{1}{k_{2}} = 1 = \frac{1}{k} + \frac{1}{k_{2}} + \frac{1}{k_{2}$ () Stale 14/1200>===12>+==13>+==15> Orthonomal states means <m In>= Sm, m Thus normalization means くサイモ=0)サイモ=の》=+++++====1 Or 161=2, C=2.eux Also  $\frac{1}{1+(t>0)} = \frac{1}{2}e^{-t(2+1/2)20t} |2\rangle$   $\frac{1}{1+2}e^{-t(3+1/2)20t} |3\rangle$   $\frac{1}{2}e^{-t(x+1/2)20t} |5\rangle.$ 

Problem 3 4) Split: Ho = p2 + Vo H. = Y ( =) Portune Vocal = 20 for - beter log for - beter log for x36 is use for for proceeding by perturbato ground - state everyn Ep= tr2 TT2 (as described by 2m (23)2, Ho Hamiltonian) ) First-order pertubation theory SEPERATIN, 14, 14, > where Ho ground-state eigenvedor 1717 is given 4, (x) = (x) 4, > = (= Sin (T(x+b))  $= \overline{P}^{\circ} \left( \frac{1}{25} \right)$ 

 $SE^{(1)}_{b} = \frac{1}{b} \int_{-b}^{b} dx \left( \frac{x}{b} \right) \cdot \cos^{2}\left( \frac{\pi x}{2b} \right)$ = 1 dy . y. cos? (= y) = 0 So  $E_1 \subseteq E_1^{(1)} = E_1^{(2)} = E_1^{(2)} = \frac{\pi^2 h}{8mh^2}$ c) Second order porturbation theory assuming we can focus on coupling to first excited state  $E_2 = \frac{T^2 + 2}{2} + \frac{1}{4_2} + \frac{1}{4$ SE(2) = 5 15 + 1 H, 1 + n > 16 n = 1 = 16 = 6  $2 = \frac{1}{2} \cdot \frac{3 \text{ mb}}{772 \text{ K2}} \left[ \text{KH} + 1 \text{H} + 2 \text{ K2} \right]^2$ We need matrix · pleur 641(4/651745>  $= \int dx(\tilde{c}) \frac{1}{5} \sin(\frac{\pi}{5}) \cos(\frac{\pi}{5})$ 2 - 17 9/2 = ( dy y sin ( TT y) cos ( TT y /2) y = <u>27</u> =  $\left(\frac{2}{\pi}\right)^2 \cdot 2 \int dz \cdot z \cdot \sin(z) \cos^2 z$ 

 $= \left(\frac{2}{\pi}\right)^{2} \cdot \frac{2}{q} \left[-\frac{3}{32\cos^{3}(2)} + \sin 2\cos^{2}(2) + 2\sin 2\right]^{2}$  $= \frac{2}{3} \left(\frac{2}{3}\right)^2$ Thus change is approximatively SE(2) S - 3 Smb2 . y2. (33)2 d) DH, H) = (Cot cos (wt); C = e Erad d) Time-dependent perturbation theory, Fermi golden rule Y1-2 = Iniky, 1.C. 2142>12 S(w-w12) where  $t_1 w_{12} = E_2^2 - E_1^2$ But we know the matrix elements already VIDZ = The (e Erad b)2. [8(2)2] S(w-wiz)

Problem 4,  $\overline{B} = B_0(1,0,0)$ Spin Hauntonian H=-1, N.B. , N= 9MB. 5 ; g+2 Thus  $H \simeq B_{NO} \simeq = (N_{O})B_{O}(1)$ Use  $\chi_{\mathbf{A}} = \begin{pmatrix} 1 \\ n \end{pmatrix} \vee \chi_{\mathbf{V}} = \begin{pmatrix} 2 \\ n \end{pmatrix}$ to denote states with spin up/dowin in z direction Then eigenvalue problem is  $B_{0} \stackrel{(0)}{=} (10) \chi_{D} = E_{n} \chi_{n}, n = 0, 1$ for Xn = a Xn + b Xu a) Othonomal Solutions are flue  $\mathcal{N}_{o} = \frac{1}{1} \begin{pmatrix} 1 \\ -1 \end{pmatrix}$  of every  $E_{o} = -B_{o} \frac{N_{e}}{N_{o}}$ het J2B = (Bo NO)/K

10 b)  $\chi(t=0) = \chi_{1} = \binom{1}{0}$  $\chi_1 = \pm (\chi_0 + \chi_1)$ Thus X(E)= = No etilst + tax e - i Dest  $= \frac{(\cos(\partial_{cB}t))}{(-2\sin(\partial_{cB}t))}$ = cospet Xn - Esin (2st) Xy. C) Pult) = 1 sin (2Bt) 2 oscillating d) Two electrons in same spahial-wave function state (but non interacting) then they must have opposite spin-directions If one is spindown at to the other is spin up at t=t, Probability for spin down at t'>t, is then given by c':  $\overline{P_{i}(t')} = P_{i}(t'-t_{i}) = 1 \sin (\mathcal{R}B(t'-t_{i}))]^{2}$ 

# Answers to exam questions August 2017

Problem 1.e) Define the concept "electronegativity".

Answer

A measure of the ability of an atom in a chemical compound to attract electrons from another atom in the compound.

**Problem 1.f)** State the requirements needed to be fulfilled by a compound to be aromatic (not in the context of having an aroma but in the context of chemical bonding). *Answer* 

*The molecule must contain:* 

- a. A continuous and cyclic conjugated system
- b. The conjugated system must be planar
- c. The number of electrons within the conjugated system must be 4n + 2, where n is an integer including 0.

Problem 5) Fluoride gas has the molecular formula F<sub>2</sub>.

a. Draw a correct Lewis structure for this compound. (1) *Answer* 

b. Construct and draw a molecular orbital energy level diagram for F<sub>2</sub> (the "interaction between the two 2p-orbitals" along the bond axis is much larger than between those perpendicular to this axis). (2)
 Answer



c. For each energy level, give a simple (omit constants) expression for the linear combination of atomic orbitals (AOs) that gives the best approximation of the corresponding molecular orbital (MO). (1)

Answer

 $\begin{array}{l} \sigma_{2p}^{*} \propto 2p_{x} - 2p_{x} \\ \pi^{*} \propto 2p_{z} - 2p_{z} \\ \text{The same for } 2p_{y} \\ \pi \propto 2p_{z} + 2p_{z} \\ \text{The same for } 2p_{y} \\ \sigma_{2p} \propto 2p_{x} + 2p_{x} \\ 2\sigma^{*} \propto 2s - 2s \\ 2\sigma \propto 2s + 2s \\ 2\sigma^{*} \propto 1s - 1s \\ 1\sigma \propto 1s + 1s \end{array}$ 

Correlate each bonding electron pair and free electron pair, if any, shown in your Lewis structure for F<sub>2</sub> with the correct linear combination of atomic orbitals (LCAOs) constructed in question c. (2)

Answer

The Lewis structure of  $F_2$  contains one  $\sigma$ -bond. The only filled bonding  $\sigma$ -orbital that is not counteracted by a filled anti-bonding  $\sigma$ -orbital is  $\sigma_{2p} \propto 2p_x + 2p_x$ . Thus, this corresponds to the  $\sigma$ -bond.

Only valence electrons are shown in Lewis structures. Thus, there are no electrons highlighted in the Lewis structure that correspond to the linear combinations of 1s.

Therefore, the other linear combinations are represented as free electron pairs in the Lewis structure of  $F_{2}$ .